



An Insight into Anticorrosive Profiling of Drugs

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Anti-Inflammatory.

Abstract

Drugs play an important role for the wellness of communities all over the world. In addition, their potential cannot be ignored for commercial point of view especially as anticorrosive agents. We have done an intensive study for drugs such as antibiotic, anti-inflammatory, antihistamine, antifungal, antibacterial, antihypertensive, antidiabetic and other categories, to explore their potential as anti-corrosive agents against various metals. These drugs have been studied in acidic environments such as HCl, H₂SO₄, H₃PO₄, and NaCl against a variety of metals including mild steel, carbon steel, aluminum, copper, iron, zinc, and their alloys. The outcome of our study revealed that drugs are equally effective as corrosion inhibitors. Furthermore, comparative studies of several drug categories revealed that antibiotic and anti-inflammatory drugs demonstrated significant potential of metal protection from corrosion and hence can be used for commercial purpose. The futuristic approach of our work is to give it an extension to the expired drugs' domain which contribute to the solid waste all over the world. Moreover, new dimension can be given to pharmaceutical and paint industries for the use of expired drugs as corrosion inhibitors.

1. Introduction

Corrosion is defined as the destruction or decadence caused by a lack of metals or alloys because of a chemical or electrochemical attack from the surrounding environment. The principal causes of metal corrosion include atmospheric, air, and water contamination, as well as the metals' conducting surfaces [1]. It is a natural process that has significant effects on metals or alloys in aqueous solutions. Corrosive substances in the fluids include water or saltwater, dissolved carbon dioxide, oxygen, and acids such as H₂CO₃, HCl, and H₂SO₄. Corrosion has become a major issue for several businesses throughout the world, particularly in the oil and gas industries. By electrochemical processes, the transportation of fluids containing crude oil, natural

gas, and caustic compounds causes rust in the inner section of the pipeline [2]. This corrosion must be prevented or managed, otherwise the special qualities of low carbon steel may be lost. The cost of scale wastes from industries in India is 26.1 billion USD, or 2.4% of total GDP [3]. The worldwide loss due to corrosion is estimated to be \$2.5 trillion (about \$7,700 per person in the US) (about \$7,700 per person in the US), or around 3.5% of global GDP (NACE March 2016) [4]. Corrosion costs have typically been expressed as a percentage of a country's Gross National Product (GNP), ranging from about 2% to 5%, depending primarily on whether indirect losses attributable to corrosion have been considered in addition to direct losses. GDP and

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GNP have been used completely arbitrarily, with no clearly established criteria. So, corrosion is continuously affected on over GDP and GNP, some historic GNP data is given in the figure 1. [5]

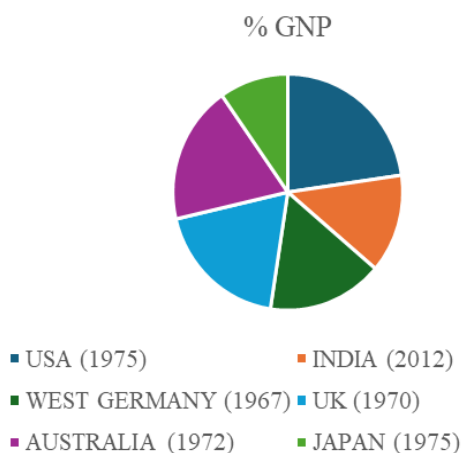


Figure 1 Some Historic Events of Corrosion Cost

In the twenty-first century, effectively reducing corrosion on metals in a range of conditions remains a difficulty. Because of the sustained economic progress and the cost-of-living efforts to reduce corrosion, various methods such as electrochemical defense and the addition of a small quantity of substances (inhibitor) that reduces the corrosion process in acidic environments were used. Inhibitors are added to the metal to protect the metal surface, resulting in the formation of stable compounds. Furthermore, numerous organic, inorganic, herbal extracts, pharmaceutical medications, ionic liquids, and synthetic chemicals were employed as inhibitors on a regular basis [6]. Because they include multiple functional groups and unsaturation as adsorption centers, pharmaceutically active substances are used as corrosion inhibitors. They are biodegradable and may be produced and processed at a low cost [7]. Antibiotics such as ampicillin, cloxacillin, flucloxacillin, and amoxicillin were among the first medications tested as corrosion inhibitors, with inhibitory efficiency reaching up to 90%. Antihypertensive medications (Enalapril maleate, Atenolol, and Etilefrine) were successfully evaluated in the same solutions [8]. The negative impact of inorganic inhibitors on environmental safety prompted researchers to shift their focus to organic chemicals or natural products. These compounds' inhibitory capacity is proportional to the availability of electrons and functional groups such as $-CO_2H$, $-CO_2R$, $-NR_2$, $-NH_2$, $-OH$, $-OR$, and $-SR$ [9]. So yet, there is no information on the direct use of expired

medications as corrosion inhibitors in the petroleum industry. Corrosion inhibitors derived from expired pharmaceuticals are still being studied to generate better inhibitors (with an inhibition efficacy of more than 99%) that can be compared to conventional inhibitors. Expired medications are unsafe to consume due to the degradation of active ingredients as well as excipients caused by physical, chemical, or microbiological influences such as humidity, pressure, temperature, bacteria, and light [10]. The structure of traditional and expired medications is comparable to that of general organic inhibitors, which include functional groups and electrons to enable their performance as corrosion inhibitors [11]. In our work, we tried to prepare a complete profiling of drugs in an arena of anticorrosion study on metals.

1.1. Useful Metals Which Can Be Protected by Corrosion Inhibitors:

There are many metals we use for daily or industrial purposes which are shown in figure 2.

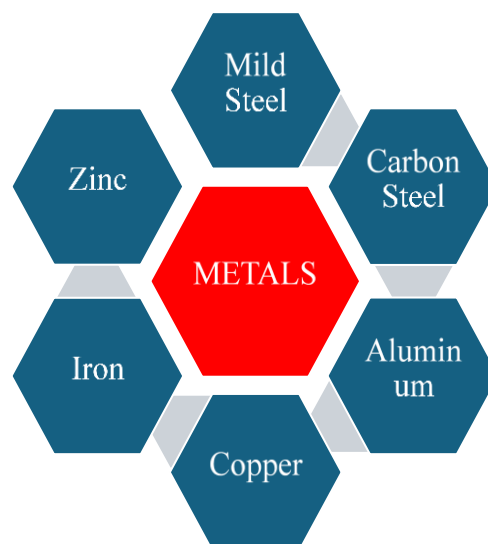


Figure 2 Useful metals for Corrosion Protection

1.2. Experimental Methods Used for Corrosion Inhibitory Study:

There are so many methods which are used to measure the corrosion inhibition efficiency such as Weight Loss, Potentio Dynamic Polarization (PDP), Galvanostatic Polarization (GAP), Electrochemical Impedance Spectroscopy (EIS), Electrochemical Frequency Modulation (EFM), and quantum analysis which is also useful. In addition, few methods are used to detect surface analysis like Scanning Electron Microscopy (SEM), Atomic Force Microscopy (AFM) and Fourier Transform Infrared spectroscopy (FTIR). Few common methods with its scope are illustrated in figure. 3

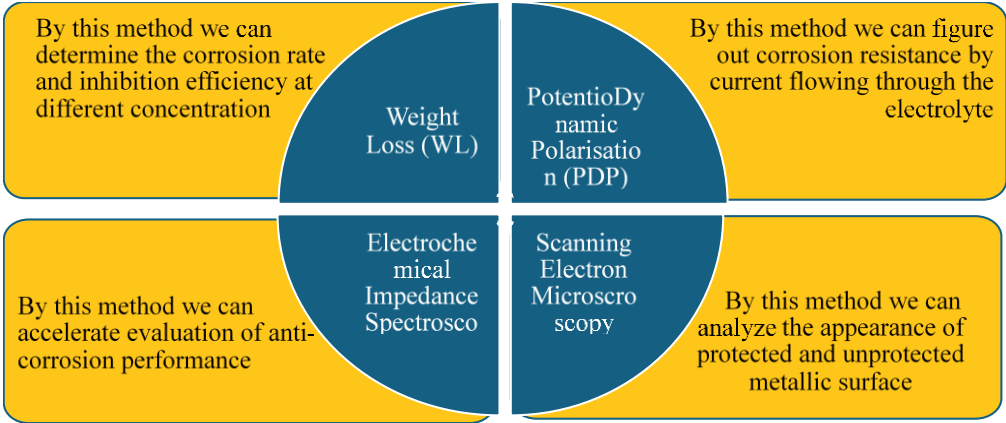


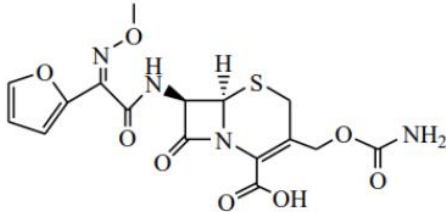
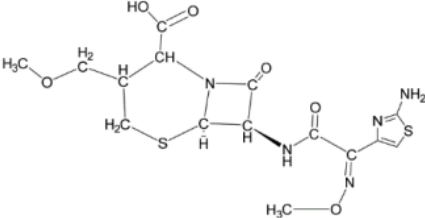
Figure 3 Useful Experimental Methods in Corrosion Inhibitory Study

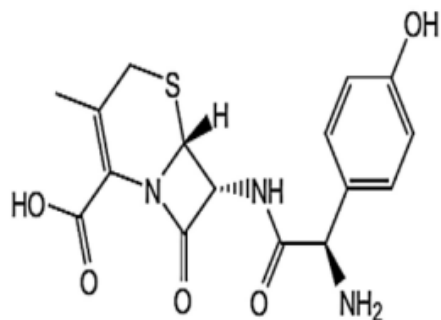
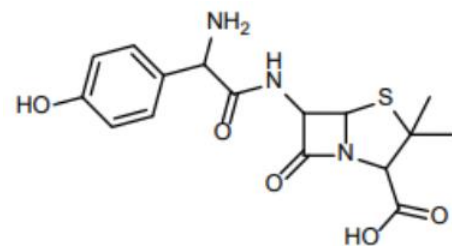
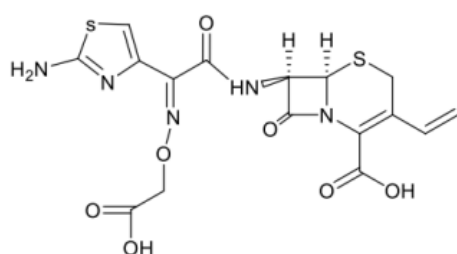
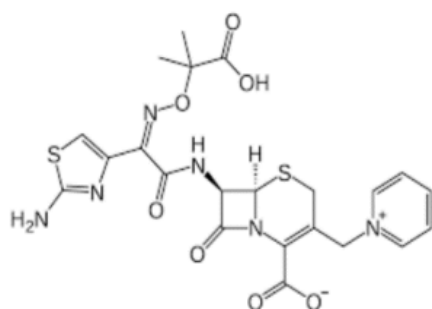
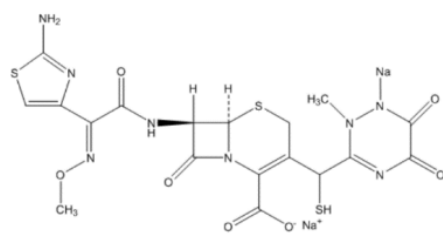
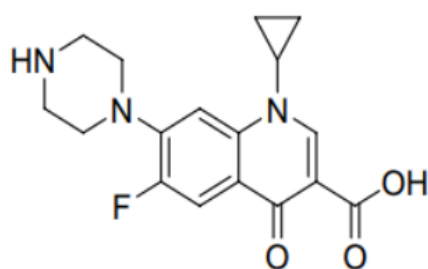
1.3. Study for Various Category Drugs Used for Corrosion Inhibition:

Our work explores different type of drugs like antibiotic, anti-inflammatory, antihistamine, antifungal, antibacterial, antidiabetic, analgesics, antiemetic, antihypertensive, anthelmintic, carbonic anhydrase, beta blocker, antiviral, laxative, skeletal

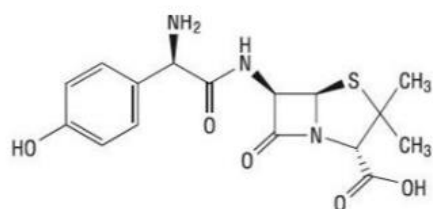
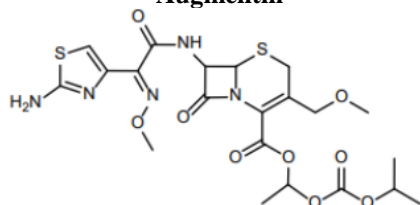
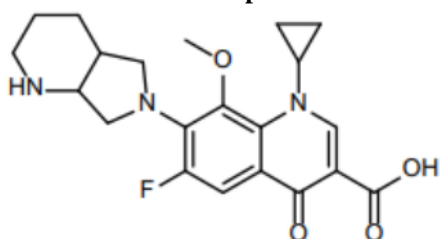
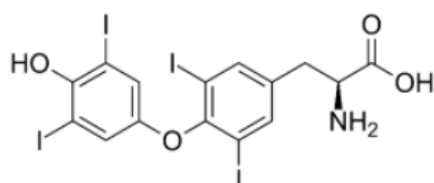
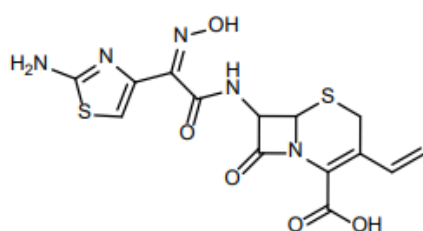
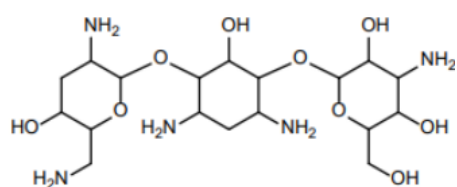
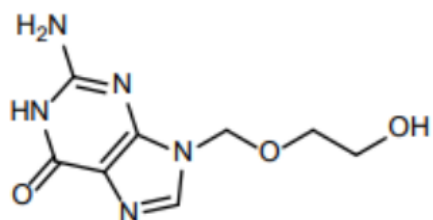
muscle relaxant, antihyperlipidemic, depressant, antagonists, sedative hypertonic, amino sugar, gastroesophageal and antiarrhythmic. Table 1 illustrates the drugs with their anti-corrosive activities against metals, in different mediums, and inhibition efficiency.

Table 1 Corrosion Analytics of Selected Category Drugs as Reported in Literature

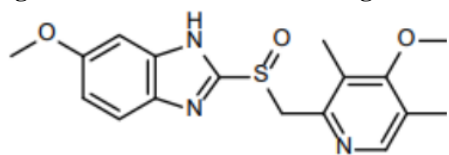
Drug category	Metal	Medium	Methods	Highest Corrosion Inhibition (IE%)	Reference No.
1. Antibiotics					
 Cefuroxime	Mild Steel	0.5M H ₂ SO ₄	WL (4.5×10 ⁻⁴ M)	94.57	[12]
	Copper	1M HNO ₃	WL (5mM)	57.50	[13]
	Aluminum	HCl	Gasometric (0.5C at 303K)	89.87	[14]
	Sabic iron	1M HCl	GAP (500 mg l ⁻¹)	96.58	[15]
	Zinc	1M H ₂ SO ₄	PDP (0.1 M)	82.88	[16]
 Cefpodoxime	Mild Steel	1N H ₂ SO ₄	EIS (200ppm)	96.52	[17]
	Copper	1M HNO ₃	Quantum at B3LYP	6-31G	[18]
	Aluminum	0.1M H ₂ SO ₄	Gasometric (0.5 gl ⁻¹)	89.10	[19]
	Cast Iron	1M HCl	WL (240ppm)	95.2	[20]
	Mild Steel	1M HCl	PDP (11.0×10 ⁻⁴ mol l ⁻¹)	96	[21]
	Aluminum 6063	0.5M H ₂ SO ₄	PDP	58.9	[22]
	Copper	1M HNO ₃	WL (2 mM)	94	[23]
	Cast Steel	1M HCl	EIS (10 ⁻² M)	96.10	[24]

**Cefadroxil****Amoxicillin****Cefixime****Ceftazidime****Ceftriaxone****Ciprofloxacin**

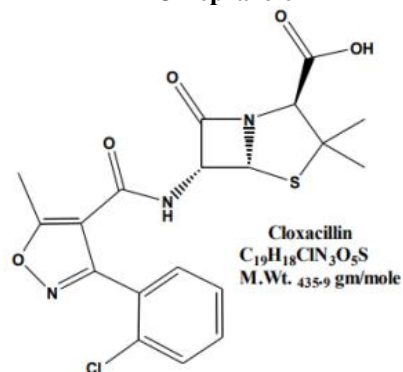
Aluminum	0.1M H ₂ SO ₄	PDP (2mM)	90.3	[25]
Mild Steel	1 M HCl	Gravimetric (1800ppm)	94.47	[26]
Carbon Steel	3.5%NaCl	PDP (250 gl ⁻¹)	91.2	[27]
Aluminum	1 M HCl	PDP (150ppm)	73	[28]
1020 Steel	NaCl	EIS (200ppm)		[29]
Sabic Iron	1 M HCl	GAP (500ppm)	94.7	[30]
Mild Steel	0.5M H ₂ SO ₄	Gravimetric (3.5×10 ⁻⁴)	96.93	[31]
Carbon Steel	H ₂ SO ₄	WL (600ppm)	90.4	[32]
Copper	1M HNO ₃	WL (2 mM)	91.07	[33]
Aluminum	HCl	ML (2 mM)	90.41	[34]
Mild Steel alloy	AgNPs	WL (3×10 ⁻³ M)	93.90	[35]
Mild Steel	HCl	PDP (250pm)	90.2	[36]
Carbon Steel	Fuel	PDP (500ppm)	97.34	[37]
Copper	1mol/L HCl	EFM (300ppm)	94.2	[38]
Aluminum	0.1M NaOH	EIS (300ppm)	78.4	[39]
Cu & Ni	Acidic	EIS	91	[40]
Mild Steel	1N HCl	PDP (400ppm)	90.95	[41]
Carbon Steel	0.5M HCl	PDP (50ppm)	93.37	[42]
Aluminum	2M HCOOH	PDP	88.09	[43]
Nickel	1M HCl	EIS (10 ⁻⁵ mol L ⁻¹)	73.09	[44]
Mild Steel	1M HCl	WL (0.009 M)	97.72	[45]
Copper	SAR (synthetic acid rain solution)	EIS (600ppm)	90.1	[46]
Bronze	Acid rain	PDP (2000ppm)	69.7	[47]

**Augmentin****Podocip****Moxifloxacin****Levothyroxine****Cefdinir****Tobramycin****Acyclovir (Exp)**

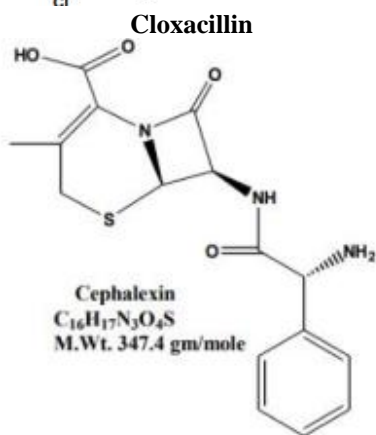
Carbon Steel alloy Copper	1N HCl	Acidimetric (300ppm)	98	[48]
	HCl	WL (500ppm)	93	[49]
α -brass	0.5 M HNO ₃	WL (300 ppm)	93.6	[50]
Carbon Steel	1M HCl	EIS (100 mg L ⁻¹)	97.3	[51]
Carbon Steel	1 M HCl	EIS (300ppm)	94.1	[52]
Aluminum	1M H ₂ SO ₄	PDP (400ppm)	86.17	[53]
Steel	2 M HCl	PDP (400ppm)	97.63	[54]
Stainless Steel	H ₂ SO ₄	ML (150ppm)	94.3	[55]
Mild Steel	1 M HCl	WL (6.32 \times 10 ⁻⁴ M at 35 °C)	97.9	[56]
Carbon Steel	2M HCl	PDP (400ppm)	92.6	[57]
Sabic Iron	HCl	GAP (500ppm)	94.47	[58]
Sabic Iron	HCl	GAP (500ppm)	95.26	[59]

**Omeprazole**

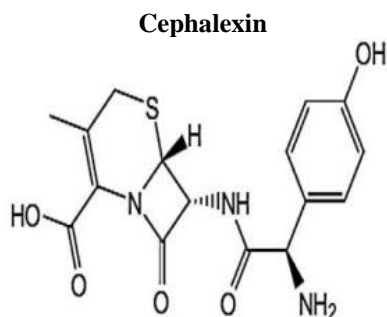
C38 Steel	1 M H ₃ PO ₄	EIS (7x 10 ⁻⁴)	99	[60]
C60 Steel	0.5 M H ₂ SO ₄	EIS (300 mgL ⁻¹)	92.52	[61]
Aluminum 6063	2M NaOH	WL (600 ppm)	87.7	[62]

**Cloxacillin**
C₁₉H₁₈ClN₃O₅S
M.Wt. 435.9 gm/mole

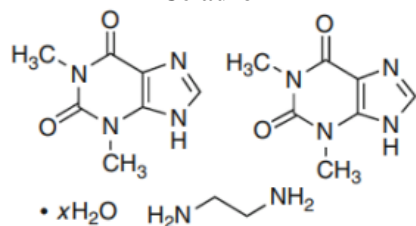
Carbon Steel	3.5% NaCl	EIS (750 mg L ⁻¹)	91.9	
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**Cephalexin**
C₁₆H₁₇N₃O₄S
M.Wt. 347.4 gm/mole

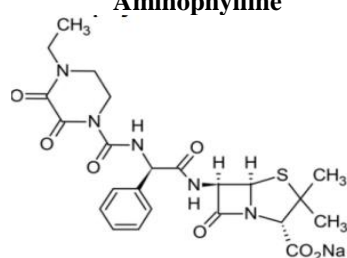
Carbon Steel	Carbon Steel	EIS (750 mg L ⁻¹)	56.25	
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**Cefadroxil**

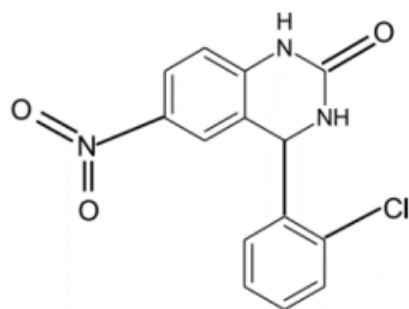
Aluminum	0.5 HNO ₃	PDP	44	[64]
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**Aminophylline**

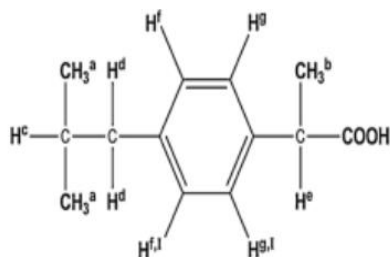
Carbon Steel	HCl	EIS (0.6 mmol L ⁻¹)	87.3	[65]
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**Piperacillin sodium**

Mild Steel	HCl	WL (500ppm)	96.12	[66]
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**Clonazepam**

Mild Steel	3.5% NaCl	3.5% NaCl	92.1	[67]
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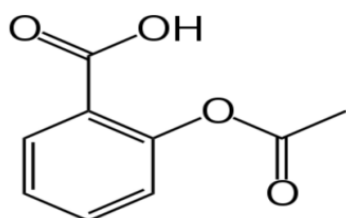
2. Anti-inflammatory**Ibuprofen**

Mild Steel	H ₂ SO ₄	PDP	63.25	[68]
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Copper	0.5M H ₂ SO ₄	EIS (5×10 ⁻³ M)	95.25	[69]
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Aluminum	0.5M H ₂ SO ₄	PDP	80.58	[70]
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Stainless steel	0.5M H ₂ SO ₄	PDP	60.59	[71]
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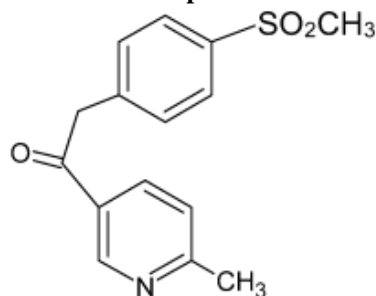
**Aspirin**

Mil Steel	0.5M H ₂ SO ₄	WL (3×10 ⁻⁵ M)	71.80	[72]
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Copper	0.5M HCl	WL (3×10 ⁻⁵ M)	67	[73]
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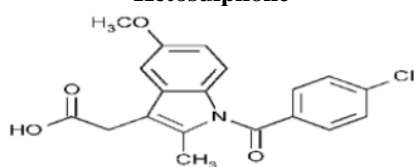
Aluminum Pure	0.5 M H ₂ S	WL	96.98	[74]
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Carbon Steel XC48	1 M HCl	EIS (5×10 ⁻³ M)	96	[75]
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**Ketosulphone**

Mild Steel	1 M HCl	WL (200 ppm)	96.61	[76]
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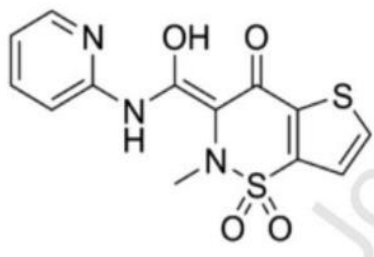
Zinc	1 M HCl	PDP (20 ppm)	52.5	[77]
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**Indomethacin**

Carbon Steel	1M HCl	WL (500 ppm)	83	[78]
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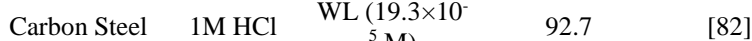
Copper	Low pH		50	[79]
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Carbon Steel	0.5 M HCl	PDP (4×10 ⁻⁴ M)	90.3	[80]
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**Tenoxicam**

Copper	1 M HNO ₃	Quantum	B3LYP	[81]
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3. Antihistamine

Clc1ccc(cc1)C2(=O)N2C3=CC=CC=C3C4=CC=C(C)C=C4CN2 · 2HCl · H

Aluminum	1M HCl	EIS (200 ppm)	95.39	[83]
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NC(=O)S(=O)(=O)c1cc(Cl)ccc1NC(=O)S(=O)(=O)N

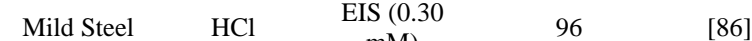
hydrochlorothiazide

Carbon Steel	1 M HCl	PDP (300 ppm)	86.9	[84]
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OCCOCCN1CCN(C1Cc2ccccc2)c3ccc(Cl)cc3

Carbon Steel	0.5 M H ₂ SO ₄	WL (400 ppm)	95	[85]
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4. Antifungal



5L X52 Steel	HCl	EIS (30 ppm)	96.6	[87]
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Aluminum	1M HCl	WL ($1 \times 10^{-4} \text{M}$ at 30°)	82.4	[88]
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TM	HCl	EIS (100 ppm)	84.19	[89]
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CC1(C)N2C(=O)N3C(=N1)C(=N3)N2C4=CC=C(C=C4)N5C6=CC=C(C=C6)N7C8=CC=C(C=C8)OCC9C(C=C(C=C9)C10=CC=C(C=C10)Cl)O[C@H](C11=CC=C(C=C11)Cl)[C@@H](C12=CC=CC=C12)N13

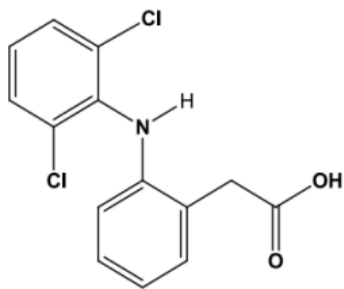
Mild Steel	HCl	WL (1.0l M at 303 K)	80.0	[90]
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5. Antibacterial

Mild Steel	1M HCl	EIS (100 nm)	98	[91]
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Aluminum	0.1 M HCl	ppm) EIS (150 nm)	99.99
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Aluminum 6061 alloy (Exp)	HCl 3.5% NaCl	ppm) EIS (150 ppm)	83.32	[92]
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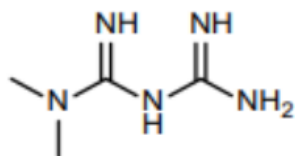
**Diclofenac sodium**

Iron

1N
H₂SO₄WL (180 ppm
at 303.15K)

42

[93]

6. Anti diabetic**Metformin**

Mild Steel

1M HCl

EIS (400
ppm)

96

[94]

C 1080 Steel
(Exp)CH₃COO
HEIS (200
ppm)

89.47

[95]

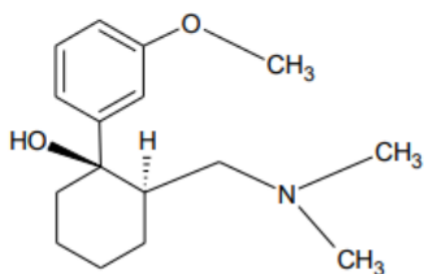
Steel

HCl

WL (400 pm)

80.01

[96]

7. Analgesics**Tramadol**

Steel

1 N
H₂SO₄WL (500
ppm)

93.2

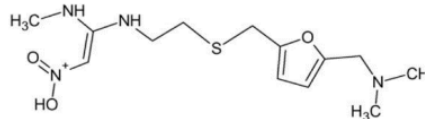
[97]

Aluminum

1.0 M
HClEIS (500
ppm)

98.4

[98]

8. Antiemetic**Ranitidine**

Mild Steel

1M HCl

EIS (400 pm)

97.55

[99]

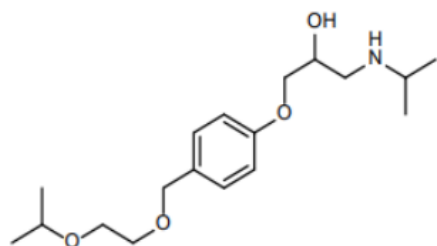
Aluminum

1M HCl

WL (333K)

88.7

[100]

9. Antihypertensive**Concor**

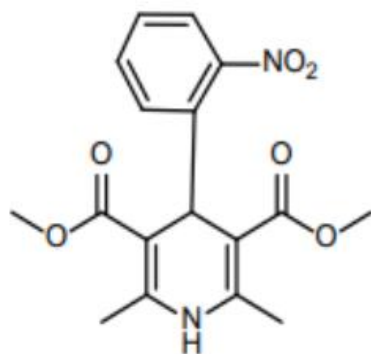
Concor

2M HCl

EIS (300
ppm)

85.8

[101]

**Nifedipine**

Mild Steel

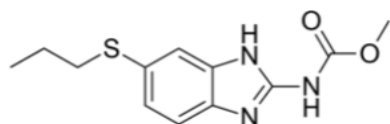
1 M HCl

EIS

95.61

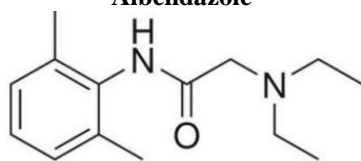
[102]

10. Anti-Helminthic



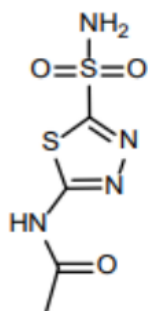
Albendazole

Mild Steel	1 M HCl	EIS (65 ppm)	86.66	[103]
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Lidocaine

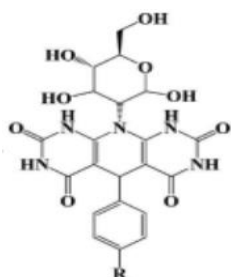
QSARARX	QSARA RX	EIS (100 ppm)	95.2	[104]
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11. Carbonic Anhydrase

Acetazolamide

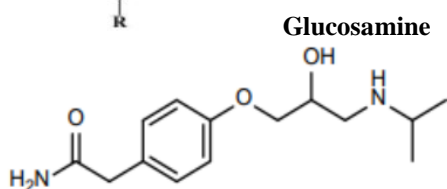
Mild Steel	HCl	EIS (500 ppm)	96.27	[105]
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Carbon Steel	1M HCl	WL (300 ppm)	86.1	[106]
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12. Beta Blocker

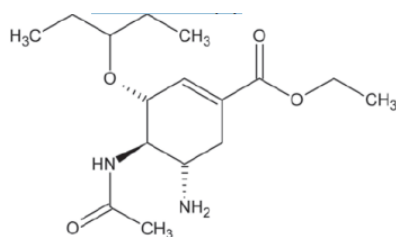
Glucosamine

Carbon Steel	0.75 mmolL ⁻¹	PDP (GS + 0.75 mmol L ⁻¹ KI)	91.2	[107]
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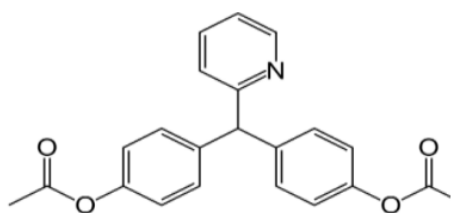
Atenolol

Mild Steel	1 M HCl	1 M HCl	93.29	[108]
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13. Antiviral

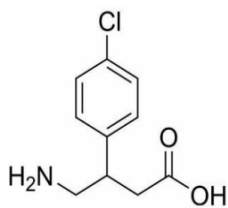
Oseltamivir

Aluminum	3M HCl	WL (2 mg L ⁻¹)	99.10	[109]
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14. Laxative

Dulcolax (DULC)

Carbon Steel	1 HCl	WL (500 ppm)	91.80	[110]
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15. Skeletal Muscle relaxant**Baclofen**

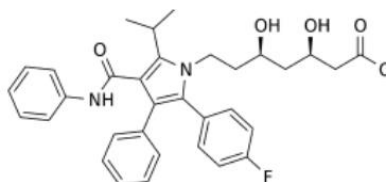
Steel

1 N HCl

Gasometric
(100 ppm)

98

[111]

16. Antihyperlipidemic**Atorvastatin**

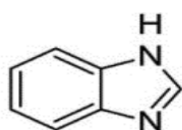
Mild Steel

1M HCl

PDP (150
ppm)

99.08

[112]

17. Depressant**Benzimidazole**

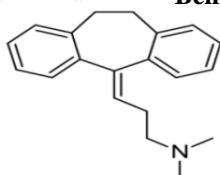
Mild Steel

1 M HCl

EIS (250
ppm)

90.4

[113]

**Amitriptyline**

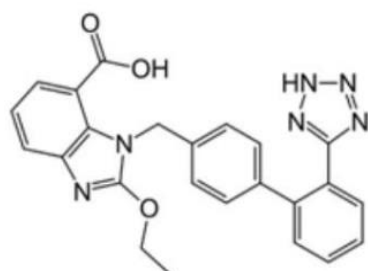
Mild Steel

3M HCl

PDP (2 mg L⁻¹)

87.57

[114]

18. Antagonist**Candesartan**

Carbon Steel

HCl

1M HCl

91.20

[115]

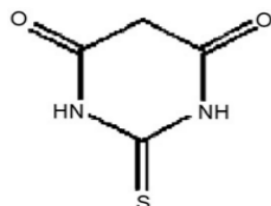
Aluminum

1M HCl

EIS (20 ppm)

92.9

[116]

19. Sedative Hypertonic**Barbiturates**

Mild Steel

HCl

PDP

97.7

[117]

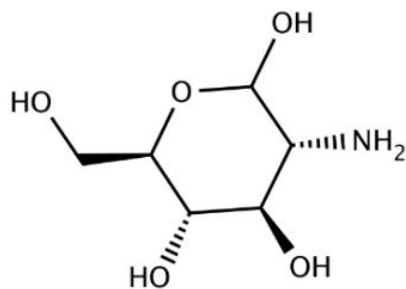
API 5L X60
steel

NaCl

PDP (75
ppm)

98.9

[118]

20. Amino Sugar**Glucosamine**

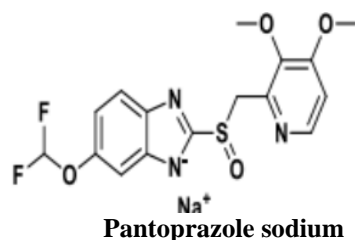
Carbon Steel

H₂SO₄PDP
(200ppm)

82.2

[119]

21. Gastroesophageal



Carbon Steel

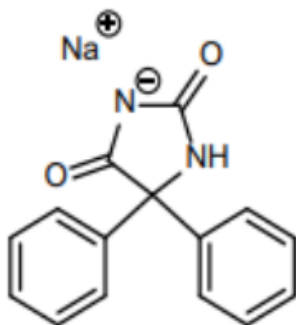
1M HCl

ML (300 ppm)

93.3

[120]

22. Antiarrhythmic



Carbon Steel

1M HCl

PDP (500 ppm)

81.78

[121]

WL= Weight Loss ML= Mass Loss

2. Result and Discussion

2.1. Comparative Anti-Corrosive Efficiency of Various Category Drugs

After studying the table 1, we plotted a graph as shown in figure 4 which shows that antibiotic and non-steroidal anti-inflammatory drugs (NSAIDs) are mostly used as corrosion inhibition purpose. Therefore, we focused primarily on antibiotics and antibiotic, and secondarily on other drugs categories in this research work.

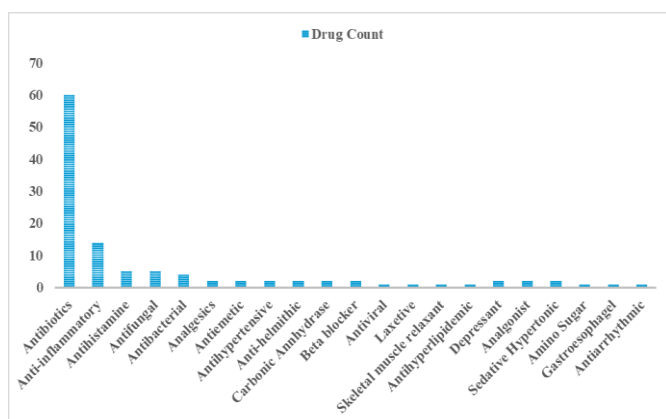


Figure 4 Anticorrosive Effect Given by Different Drug Categories

2.2. Study for Antibiotic Drugs' Anticorrosive Efficiency Against Metals

For this study we analyzed the percentage (%) inhibitory efficiency of antibiotic drugs on number of metals as given in table-2.

Table 2 Anticorrosive Action of Antibiotic Drugs

on Metals			
Metals	Antibiotic drugs	% inhibitor efficiency	Avg. of % inhibitor efficiency
Mild steel	Cefuroxime	94.57	95.00
	Cefpodoxime	96.52	
	Cefadroxil	96	
	Amoxicillin	94.47	
	Cefixime	96.93	
	Cefixime	93.90	
	Ceftazidime	90.2	
	Ceftriaxone	90.95	
	Ciprofloxacin	97.72	
	Cefdinir	97.9	
	Omeprazole	99	
	Omeprazole	92.52	
	Piperacillin sodium	96.12	
	Clonazepam	92.1	
Carbon steel	Amoxicillin	91.2	89.97
	Cefixime	90.4	
	Ceftazidime	97.34	
	Ceftriaxone	93.37	
	Augmentin	98	
	Podocip	97.3	
	Moxifloxacin	94.1	
	Tobramycin	92.6	
	Cloxacillin	91.9	
	Cephalexin	56.25	
	Aminophylline	87.3	

Aluminum	Cefuroxime	89.87	79.63
	Cefpodoxime	89.10	
	Cefadroxil	58.9	
	Cefadroxil	90.3	
	Amoxicillin	73	
	Cefixime	90.41	
	Ceftazidime	78.4	
	Ceftriaxone	88.09	
	Moxifloxacin	86.17	
	Omeprazole	87.7	
Copper	Cefadroxil	44	85.39
	Cefuroxime	57.50	
	Cefadroxil	94	
	Cefixime	91.07	
	Ceftazidime	94.2	
	Ciprofloxacin	90.1	
	Ciprofloxacin	69.7	
	Augmentin	93	
Iron	Augmentin	93.6	94.00
	Cefuroxime	96.58	
	Amoxicillin	94.7	
	Acyclovir (Exp)	94.47	
Zn & Ni	Omeprazole	95.26	82.23
	Cefuroxime	82.88	
	Ceftazidime	91	

2.3. Graphical Representation of Antibiotic Drugs' Efficiency Against Metals

Figure 5 indicates that anticorrosive action of antibiotic drug is highest on mild steel and iron whereas it is lowest on aluminum.

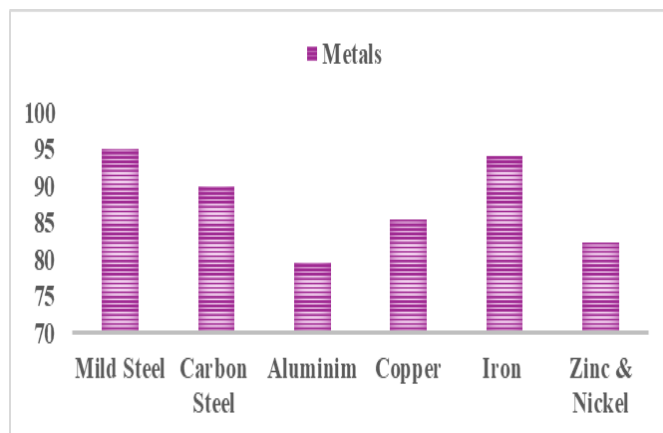


Figure 5 Average Inhibition Efficiency of Antibiotic Drugs

2.4. Study for Anti-Inflammatory Drugs' Anticorrosive Efficiency Against Metals

For this study we analyzed the percentage (%) inhibitory efficiency of anti-inflammatory drugs on number of metals as given in table-3.

Table 3 Anticorrosive Action of Anti-Inflammatory Drugs on Metals

Metals	Anti-inflammatory drugs	% inhibitory efficiency	Avg. of % inhibitory efficiency
Mild steel	Ibuprofen	60.59	73.06
	Aspirin	71.80	
	Keto sulphone	96.61	
Carbon steel	Aspirin	96	89.76
	Indomethacin	83	
	Tenoxicam	90.3	
Aluminum	Ibuprofen	80.58	88.78
	Aspirin	96.98	
Copper	Ibuprofen	95.25	70.75
	Aspirin	67	
	Indomethacin	50	
Iron	Keto sulphone	52.5	52.50

2.5. Graphical Representation of Anti-Inflammatory Drugs' Efficiency Against Metals

Figure 6 indicates that anticorrosive action of anti-inflammatory drug is highest on carbon steel and aluminum whereas it is lowest on iron.

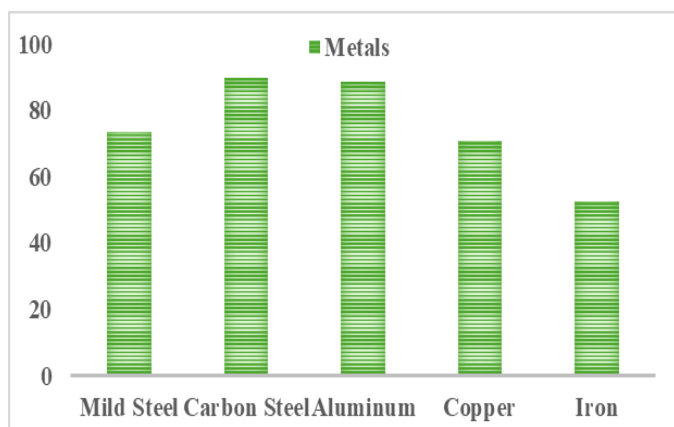


Figure 6 Average Inhibition Efficiency of Anti-Inflammatory Drugs

2.6. Study for Other Categories Drugs for Anticorrosive Efficiency Against Metals

For this study we analyzed the percentage (%) inhibitory efficiency of miscellaneous category drugs on number of metals as given in table-4.

Table 4 Anticorrosive Action of Other Category Drugs on Metals

Metals	Miscellaneous category drugs	% inhibitory efficiency	Avg. of % inhibitory efficiency
Mild steel	Fluconazole	96	85.57
	Itraconazole	80.0	
	Diclofenac sodium	98	
	Metformin	96	
	Metformin	80.01	
	Tramadol	93.2	
	Ranitidine	97.55	
	Nifedipine	95.61	
	Albendazole	86.66	
	Acetazolamide	96.27	
	Atenolol	93.29	
	Baclofen	98	
	Atorvastatin	99.08	
	Benzimidazole	90.4	
Carbon steel	Amitriptyline	87.57	88.38
	Barbiturates	97.7	
	Acetazolamide	86.1	
	Glucosamine	91.2	
	Dulcolax	91.80	

	(DULC)		
	Candesartan	91.20	
	Glucosamine	82.2	
	Pantoprazole sodium	93.3	
	Phenytoin sodium	81.78	
Aluminum	Fluconazole	82.4	93.58
	Diclofenac sodium	99.99	
	Tramadol	98.4	
	Ranitidine	88.7	
	Oseltamivir	99.10	
	Candesartan	92.9	
Iron	Diclofenac sodium	42	42.00

2.7. Graphical Representation of Other Categories Drugs for Inhibitory Efficiency Against Metals

Figure 7 indicates that anticorrosive action of miscellaneous category drugs is highest on aluminum followed by carbon steel, mild steel and then iron.

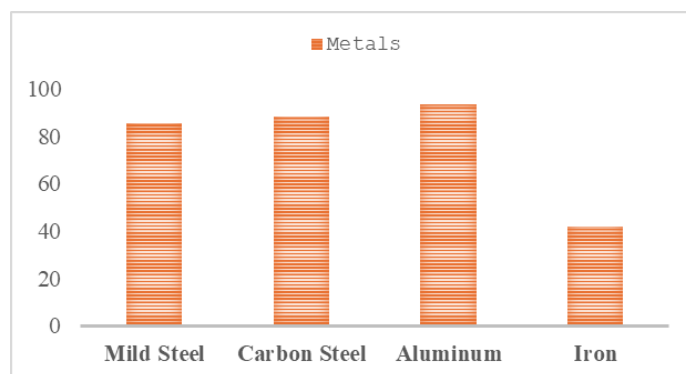


Figure 7 Average Inhibition Efficiency of Other Drugs Categories

Conclusion

Our work reveals that drugs in have potent anticorrosive efficiency on metals like mild steel, carbon steel, copper, aluminum, zinc, iron, etc. which have so many industrial as well as local applications. These metals can be protected by using corrosion inhibitors such as organic, inorganic, plant based, etc. Our work explored the anticorrosive potential of drugs, furthermore the discussion through various graphs revealed that antibiotic drugs emerged as leader followed by anti-inflammatory in all categories of drugs with good corrosion inhibition efficiency. If we study average corrosion inhibition efficiency rate especially in antibiotic drugs, it demonstrated that antibiotics have the greatest effects on mild steel and iron, and the least effects on

aluminum. Similar study for average corrosion inhibition efficiency rate in case of anti-inflammatory drugs, it showed the highest effect on carbon steel and aluminum content, followed by mild steel, copper, and iron. Other categories drugs showed that the metal with the highest corrosion inhibition efficiency rate is aluminum, followed by carbon steel, mild steel, iron, and finally steel. This work paves the way that for anticorrosive profiling for expired drugs. This futuristic study can give a tool to pharmaceutical industries for addressing the solid waste management generated by drugs all over the world, and to paint industries for preparing affordable emulsion with the help of expired medicines.

Declaration of Competing Interest: The authors declare that there is no conflict of interest.

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References

- [1]. Abd, A.N., Nasif, E.S., 2023. Corrosion inhibition of aluminum pure in acidic media using expired drugs aspirin. AIP Conf. Proc. 2475, 040003. <https://doi.org/10.1063/5.0103398>.
- [2]. Baari, M.J., Sabandar, C.W., 2021. A Review on Expired Drug-Based Corrosion Inhibitors: Chemical Composition, Structural Effects, Inhibition Mechanism, Current Challenges, and Future Prospects. Indones. J. Chem. 21(5), 1316-1336.
- [3]. Srinivasan, P., 2023. Anticorrosive concert of expired drug acts as novel and potent inhibitor for deterioration of Low carbon steel in acidic Environment. <https://doi.org/10.21203/rs.3.rs-2470589/v1>
- [4]. Gupta, N. K., Gopal, C. S. A., Srivastava, V., Quraishi M. A., 2017. Application of expired drugs in corrosion inhibition of mild steel. Int. J. Pharm. Chem. Anal, 4(1), 8-12.
- [5]. Biezma, M. V., San Cristobal, J. R., 2005. Methodology to study cost of corrosion. Corros. Eng. Sci. Technol. 40(4), 344-352.
- [6]. Gunasekaran, M., Srinivasan, P., Zulfareen, N., Venugopal, T., (2023). A case study on expired drugs: The potential corrosion inhibitory activity of expired labetalol drug in 1M HCl for plain carbon steel. Indian J. Chem. Technol. 180-186.
- [7]. TANWER, S., Shukla, S. K., 2023. Cefuroxime: A potential corrosion inhibitor for mild steel in sulphuric acid medium. Prog. Color. Color. Coat. 16(2), 125-138.
- [8]. Abeng, F.E., Ikpi, M.E., Ushie, O.A., Anadebe, V.C., Nyong, B.E., Obeten, M.E., Okafor, N.A., Chukwuike, V.I. Nkom, P.Y., 2021. Insight into corrosion inhibition mechanism of carbon steel in 2 M HCl electrolyte by eco-friendly based pharmaceutical drugs. Chem. Data Collect. 34, 100722.
- [9]. Agrawal, Y. K., Talati, J. D., Shah, M. D., Desai, M. N., Shah, N. K. (2004). Schiff bases of ethylenediamine as corrosion inhibitors of zinc in sulphuric acid. Corros. Sci. 46(3), 633-651.
- [10]. Attia, E. M., 2015. Expired Farcolin drugs as corrosion inhibitor for carbon steel in 1 M HCl solution. J. of Basic App. Chem. 5(1), 1-15.
- [11]. Pathak, R. K., Mishra, P., 2016. Drugs as corrosion inhibitors: a review. Int. J. Sci. Res, 5(4), 671-677.
- [12]. TANWER, S., Shukla, S. K. 2023. Cefuroxime: A potential corrosion inhibitor for mild steel in sulphuric acid medium. Prog. Color. Color. Coat. 16(2), 125-138.
- [13]. Ouedraogo, A., Diki, N. Y. S., Bohoussou, K. V., Soro, D., Trokourey, A., 2018. Copper corrosion inhibition by cefuroxime drug in 1 M nitric acid. Chem. Sci. Rev. Lett. 7, 427-437.
- [14]. Ameh, P. O., Sani, U. M., 2015. Cefuroxime axetil: A commercially available pro-drug as corrosion inhibitor for aluminum in hydrochloric acid solution. J. Heterocycl. Chem. 101, 1-6.
- [15]. Abdallah, M., Fawzy, A., Bahir, A. A., 2020. Expired amoxicillin and cefuroxime drugs as efficient anticorrosives for Sabc iron in 1.0 M hydrochloric acid solution. Chem. Eng. Commun. 209(2), 158-170.
- [16]. Sani, U. M., Ameh, P. O., 2015. Evaluation of cefuroxime axetil drug as acid corrosion inhibitor for zinc using polarization and weight loss technique. Int. J. Clin. Chem. Lab. Med. 1, 9-18.
- [17]. Sheetal, S. T., Singh, A. K., 2021. Corrosion inhibiting potential of Cefpodoxime-an antibiotic on mild steel in sulphuric acid. ICP13.

- [18]. Ouedraogo, A., Irie, B. I. W., Hippolyte, C. N., Trokourey, A., 2018. Copper corrosion inhibition by cefpodoxime drug in 1M Nitric Acid: experimental and DFT approaches. *Int. J. Innov. Stud.* 24(3), 1299-1311.
- [19]. SANI, U. M., AMEH, P. O., 2015. Inhibiting effect of Cefpodoxime proxetil prodrug on aluminum corrosion in 0.1 M sulphuric acid. *J. App. Phys. Sci. Int.* 5(1), 16-22.
- [20]. Rajeswari, V., Devarayan, K., Viswanathamurthi, P., 2017. Expired pharmaceutical compounds as potential inhibitors for cast iron corrosion in acidic medium. *Res. Chem. Intermed.* 43, 3893-3913.
- [21]. Shukla, S. K., Quraishi, M. A., Ebenso, E. E., 2011. Adsorption and corrosion inhibition properties of cefadroxil on mild steel in hydrochloric acid. *Int. J. Electrochem. Sci.* 6(7), 2912-2931.
- [22]. Fayomi, O. S. I., Akande, I. G., Popoola, A. P. I., Molifi, H., 2019. Potentiodynamic polarization studies of Cefadroxil and Dicloxacillin drugs on the corrosion susceptibility of aluminium AA6063 in 0.5 M nitric acid. *J. Mater. Res. Techno.* 8(3), 3088-3096.
- [23]. Yao, J. S., Trokourey, A., 2019. Thermodynamic and DFT studies on the behavior of cefadroxil drug as effective corrosion inhibitor of copper in one molar nitric acid medium. *J. Mater. Environ. Sci.* 10, 926-38.
- [24]. El Aloua, A., Oubahou, M., El Bouari, A., Tanane, O., 2023. Expired amoxicillin as an eco-friendly corrosion inhibitor for cast steel in sulfuric acid environment: electrochemical, surface and thermodynamic studies. *J. Solid State Electrochem.* 1-15.
- [25]. Diki, N. Y. S., Bohoussou, K. V., Kone, M. G. R., Ouedraogo, A., Trokourey, A., 2018. Cefadroxil drug as corrosion inhibitor for aluminum in 1 M HCl medium: experimental and theoretical studies. *IOSR J. App. Chem.* 11(4), 24-36.
- [26]. Fadila, B., Sihem, A., Sameh, A., Kardas, G., 2019. A study on the inhibition effect of expired amoxicillin on mild steel corrosion in 1N HCl. *Mater. Res. Express* 6(4), 046419.
- [27]. Jawad, A., Muallah, S. K., Hammud, K. K., 2023. Expired Colxacillin, Amoxicillin, and Ceflaxin Drugs as Inhibitors for Low Carbon Steel Corrosion in Sodium Chloride. *Al-Khwarizmi Eng. J.* 19(1), 81-98.
- [28]. Nataraja, G., Praveen, B. M., Kumar, S., Pruthviraj, R. D., 2021. Corrosion Inhibition Property of Amoxicillin Drug for Al-2014 Alloy in 1M HCl Solution. <https://doi.org/10.21203/rs.3.rs-445562/v1>.
- [29]. Scholant, M. C., Coutinho, E. F., Dias, S. P., Azambuja, D. S., Silva, S. N., Tamborim, S. M., 2015. Corrosion inhibition of AISI 1020 steel based on tungstate anion and amoxicillin as corrosion inhibitors in 0.05 mol l⁻¹ NaCl solution or inserted into cellulose acetate films. *Surf. Interface Anal.* 47(2), 192-197.
- [30]. Abdallah, M., Fawzy, A., Bahir, A. A., 2020. Expired amoxicillin and cefuroxime drugs as efficient anticorrosives for Sabc iron in 1.0 M hydrochloric acid solution. *Chem. Eng. Commun.* 209(2), 158-170.
- [31]. Tanwer, S., Shukla, S. K., 2022. Corrosion inhibition activity of cefixime on mild steel surface in aqueous sulphuric acid. *Prog. Color. Coat.* 15(3), 243-255.
- [32]. Fouda, A. S., Abdallah, A., Yousef, M., 2014. Corrosion inhibition and adsorption properties of cefixime on carbon steel in acidic medium. *Chem. Sci. Rev. Lett.* 3(11S), 130-143.
- [33]. Coulibaly, N. H., Kambiré, O., Trokourey, A., 2021. Experimental and theoretical investigations on copper corrosion inhibition by cefixime drug in 1M HNO₃ solution. *J. Mater. Sci. Chem. Eng.* 9(5), 11-28.
- [34]. Gbassi, G. K., Ouedraogo, A., Berte, M., Trokourey, A., 2018. Aluminum corrosion inhibition by cefixime drug: experimental and DFT studies. *J. Electrochem. Sci. Eng.* 8(4), 303-320.
- [35]. Siddiqua, Q.S., Tarafder, C., Daizy, M., Ahommed, M.S., Islam, M.R., Nahar, N., Hasan, M.R., Al-Mamun, M.R., Aly, M.A.S. Khan, M.Z.H., 2019. Application of Stabilized Cefixime-Agnps-Go Thin Films as Corrosion Inhibitors for Mild Steel Alloy. <http://purkh.com/index.php/tochem>.
- [36]. Singh, A. K., Shukla, S. K., Quraishi, M. A., 2011. Corrosion behavior of mild steel in sulphuric acid solution in presence of ceftazidime. *Int. J. Electrochem. Sci.* 6(11), 5802-5814.
- [37]. Anaee, R. A., Hassan, M. J. M., 2023. Experimental Study to Inhibit Carbon Steel in

- Simulated Fuel Medium by Ceftazidime Drug. *J. Surv. Fish. Sci.* 10(3S), 2171-2185.
- [38]. Kamel, M. M., Mohsen, Q., Anwar, Z. M., Sherif, M. A., 2021. An expired ceftazidime antibiotic as an inhibitor for disintegration of copper metal in pickling HCl media. *J. Mater. Res. Technol.* 11, 875-886.
- [39]. Mohamedien, H. A., Kamal, S. M., El-Deen, A. G., Taha, M., El-Deeb, M. M., 2022. Electrochemical and computational estimations of cephalosporin drugs as eco-friendly and efficient corrosion inhibitors for aluminum in alkaline solution. *Sci. Rep.* 12(1), 13333.
- [40]. Duca, D. A., Dan, M. L., Vaszilcsin, N., 2021. Recycling of Expired Cefamil Drug as Additive in the Copper and Nickel Electrodeposition from Acid Baths. *Int. J. Environ. Res. Public Health.* 18(18), 9476.
- [41]. Shukla, S. K., Quraishi, M. A., 2009. Ceftriaxone: a novel corrosion inhibitor for mild steel in hydrochloric acid. *J. Appl. Electrochem.* 39, 1517-1523.
- [42]. Hadi, Z., 2013. Evaluation of the Ceftriaxone as Corrosion Inhibitor for Carbon Steel Alloy in 0.5 M of Hydrochloric Acid. *Int. J. Electrochem. Sci.* 8(3), 3105-3120.
- [43]. Abd El Wanees, S., Alahmdi, M. I., Hathoot, A. A., Shaltoot, S., 2015. Ceftriaxone as an Inhibitor for Corrosion of Al in Formic Acid Solutions. *ISSN 2224-7467*.
- [44]. Duca, D. A., Dan, M. L., Vaszilcsin, N., 2018. Ceftriaxone as corrosion inhibitor for nickel in acid solutions. *Adv. Eng. Forum.* 27, 74-82.
- [45]. Abdullahi, M., Usman, A. K., Sani, A. M., Danazumi, K., Umar, A. U., Yusuf, J., 2022. Investigation of some expired antibiotic drugs: Effect on the corrosion inhibition of mild steel in 0.1 M HCl medium via experimental and molecular dynamics simulation. *Chem. Rev. Lett.*, 5(4), 217-225.
- [46]. Montaser, A. A., El-Mahdy, M. S., Mahmoud, E. E., Fouda, A. S., 2024. Recycling of expired ciprofloxacin in synthetic acid rain (SAR) solution as a green corrosion inhibitor for copper: a theoretical and experimental evaluation. *J. Appl. Electrochem.* 54(2), 439-456.
- [47]. Rotaru, I., Varvara, S., Gaina, L., Muresan, L. M., 2014. Antibacterial drugs as corrosion inhibitors for bronze surfaces in acidic solutions. *Appl. surf. sci.* 321, 188-196.
- [48]. Abdel Hameed, R., Essa, A., Mohamed, D., Abdallah, M., Aljohani, M., Al-Mhyawi, S., Soluman, M. and Arafa, E.I., 2022. Evaluation of expired augmentine drugs as corrosion inhibitor for carbon steel alloy in 1.0 n hcl acidic environment using analytical techniques. *Egypt. J. Chem.* 65(4) 735-745.
- [49]. Ahmed, Z. W., Qasem, M. A., 2023. Protective Effect of Expired Drug on Corrosion of Copper in Acidic Medium. *Wasit. J. for Pure Sci.* 2(2), 215-230.
- [50]. Fouda, A. A. S., Rashwan, S. M., El, A. A. N. F., Ramadan, N. H., 2016. Unused augmentin drug as save corrosion inhibitor for α -brass in nitric acid solution. *Zašt. Mater.* 57(2), 326-338.
- [51]. Dohare, P., Chauhan, D. S., Quraishi, M. A., 2018. Expired Podocip drug as potential corrosion inhibitor for carbon steel in acid chloride solution. *Int. J of corros. scale Inhib.*
- [52]. Fouda, A. S., Shalabi, K., & E-Hossiany, A., 2016. Moxifloxacin antibiotic as green corrosion inhibitor for carbon steel in 1 M HCl. *J. Bio. Tribo-Corros.* 2, 1-13.
- [53]. Nathiya, R. S., Perumal, S., Murugesan, V., Raj, V., 2018. Expired drugs: Environmentally safe inhibitors for aluminium corrosion in 1 MH 2 SO 4. *J. Bio. Tribo-Corros.* 4, 1-13.
- [54]. Ikpi, M. E., Abeng, F. E., Obono, O. E., 2017. Adsorption and Thermodynamic Studies for Corrosion Inhibition of API 5L X-52 Steel in 2 M HCl Solution by Moxifloxacin. *World. News. Nat. Sciences.* (9), 52-61.
- [55]. Motawea, M. M., 2021. Corrosion inhibition effect of expired levothyroxine drug on stainless steel 304L in 0.5 M H₂SO₄ solution. *Int. J. Electrochem. Sci.* 16(2), 21021.
- [56]. Singh, A. K., Chugh, B., Saha, S. K., Banerjee, P., Ebenso, E. E., Thakur, S., Pani, B., 2019. Evaluation of anti-corrosion performance of an expired semi synthetic antibiotic cefdinir for mild steel in 1 M HCl medium: An experimental and theoretical study. *Results Phys.* 14, 102383.
- [57]. Abeng, F. E., Anadebe, V. C., Idim, V. D., Edima, M. M., 2020. Anti-corrosion behavior of expired tobramycin drug on carbon steel in acidic medium. *S. Afr. J. Chem.* 73(1), 125-130.

- [58]. Abdallah, M., Fawzy, A., Al Bahir, A., 2020. The effect of expired acyclovir and omeprazole drugs on the inhibition of Sabic iron corrosion in HCl solution. *Int. J. Electrochem. Sci.* 15(5), 4739-4753.
- [59]. Abdallah, M., Fawzy, A., Al Bahir, A., 2020. The effect of expired acyclovir and omeprazole drugs on the inhibition of Sabic iron corrosion in HCl solution. *Int. J. Electrochem. Sci.* 15(5), 4739-4753.
- [60]. El Hamdouni, Y., Bouhlal, F., Kouri, H., Chellouli, M., Benmessaoud, M., Dahrouch, A., Labjar, N. El Hajjaji, S., 2020. Use of omeprazole as inhibitor for C38 steel corrosion in 1.0 M H₃PO₄ medium. *J. Fail. Anal. Preven.* 20(2), 563-571.
- [61]. Liu, Y., Du, W., Yao, X., Liu, C., Luo, X., Guo, L., Guo, C., 2022. Electrochemical and Theoretical Study of Corrosion Inhibition on X60 Steel in H₂SO₄ Solution by Omeprazole. *Int. J. Electrochem. Sci.* 17(5), 220516.
- [62]. Abood, H. A., 2011. The study of the inhibitory properties of Omeprazole on the corrosion of Aluminum 6063 in alkaline media. *basrah J. Sci.* 28, 74-93.
- [63]. Jawad, A., Muallah, S. K., Hammud, K. K., 2023. Expired Colxacillin, Amoxicillin, and Ceflaxin Drugs as Inhibitors for Low Carbon Steel Corrosion in Sodium Chloride. *Al-Khwarizmi Eng. J.* 19(1), 81-98.
- [64]. Fayomi, O. S. I., Akande, I. G., Popoola, A. P. I., Molifi, H., 2019. Potentiodynamic polarization studies of Cefadroxil and Dicloxacillin drugs on the corrosion susceptibility of aluminium AA6063 in 0.5 M nitric acid. *J. Mater. Res. Technol.* 8(3), 3088-3096.
- [65]. Samide, A., Tutunaru, B., Ionescu, C., Rotaru, P., Simoiu, L. 2014. Aminophylline: thermal characterization and its inhibitory properties for the carbon steel corrosion in acidic environment. *J. Therm. Anal. Calorim.* 118, 631-639.
- [66]. Reza, I., Ahmad, E., Kareem, F., 2012. Corrosion inhibition mechanism of Piperacillin Sodium for mild steel protection in acidic media. *Afinidad.* 69(557).
- [67]. Abeng, F.E., Ita, B.I., Anadebe, V.C., Chukwuike, V.I., Etiowo, K.M., Nkom, P.Y., Ekerenam, O.O., Iroha, N.B. Ikot, I.J., 2023. Multidimensional insight into the corrosion mitigation of clonazepam drug molecule on mild steel in chloride environment: empirical and computer simulation explorations. *Results Eng.* 17, 100924.
- [68]. Fajobi, M. A., Fayomi, O. S. I., Akande, I. G., Odunlami, O. A., 2019. Inhibitive performance of ibuprofen drug on mild steel in 0.5 M of H₂SO₄ acid. *J. Bio. Tribo-Corros.* 5, 1-5.7
- [69]. El Harrari, S., Salim, A., Takky, D., Naimi, Y., 2023. Corrosion inhibition effect of expired ibuprofen drug on copper in sulfuric acid solution. *J. Electrochem. Sci. Eng.* 13(6), 1005-1013.
- [70]. Fajobi, M. A., Fayomi, O. S. I., Akande, G., Odunlami, O. A., Oluwole, O. O., 2021. Evaluation of the Inhibitive Effect of Ibuprofen Drug on the Acidic Corrosion of Aluminum 6063 Alloy. *Key Eng. Mat.* 886, 133-142.
- [71]. Sanni, O., Fayomi, O. S. I., Popoola, A. P. I., Agboola, O., 2021. Data on the effect of Ibuprofen drug derivative on 430T1 stainless steel in acid solutions. *IOP Conf. Ser.: Materials Sci. Eng.* 1107 012074.
- [72]. Kushwah, R., Pathak, R. K., 2014. Inhibition of mild steel corrosion in 0.5 M sulphuric acid solution by aspirin drug. *Int. J. Emerg. Technol. Adv. Eng.* 4(7), 880-884.
- [73]. Kayadibi, F., Sagdinc, S., Zor, S., 2020. Theoretical and experimental study of the acid corrosion inhibition of copper by aspirin (acetylsalicylic acid). *Prot. Met. Phys. Chem. Surf.* 56(1), 202-213.
- [74]. Abd, A. N., Nasif, E. S., 2023. Corrosion inhibition of aluminum pure in acidic media using expired drugs aspirin. *AIP Conf. Proc.* 2475,1.
- [75]. Merimi, C., Hammouti, B., Zaidi, K., Hafez, B., Elmsellem, H., Touzani, R., Kaya, S., 2023. Acetylsalicylic acid as an environmentally friendly corrosion inhibitor for carbon steel XC48 in chloride environment. *J. Mol. struct.* 1278, 134883.
- [76]. Narayana, H., Chander, T. H., 2014. Ketosulfone Drug as a Green Corrosion Inhibitor for Mild Steel in Acidic Medium. *10.1021/ie500232g*
- [77]. Hebbar, N., Praveen, B. M., Prasanna, B. M., Venkatesha, T. V., 2015. Corrosion inhibition behavior of Ketosulfone for Zinc in acidic medium. *J. Fund. Appl. Sci.* 7(2), 271-289.

- [78]. Hameed, R. A., Ismail, E. A., Al-Shafey, H. I., Abbas, M. A., 2020. Expired indomethacin therapeutics as corrosion inhibitors for carbon steel in 1.0 M hydrochloric acid media. *J. Bio-Tribo-Corros*, 6, 114.
- [79]. Xue, H., Xu, N., Zhang, C., 1998. Corrosion behavior of copper in a copper bearing intrauterine device in the presence of indomethacin. *Contraception*, 57(1), 49-53.
- [80]. Elabbasy, H. M., Gadow, H. S., 2021. Study the effect of expired tenoxicam on the inhibition of carbon steel corrosion in a solution of hydrochloric acid. *J. Mol. Liq.* 321, 114918.
- [81]. Donatien, E.A., Amara, B., Hadja, T., Eric, A., Rodrigue, K.A., Kalifa, M., Ahemou, D., Paulin, N. Benjamin, Y., 2023. Inhibition effect of Tenoxicam on copper corrosion in HNO₃: Experimental study and DFT. *Am. J. Mater. Sci.* 11(1), pp.7-15.
- [82]. Eid, S., 2021. Expired Desloratidine drug as inhibitor for corrosion of carbon steel pipeline in hydrochloric acid solution. *Int. J. Electrochem. Sci.* 16(1), 150852.
- [83]. Ishwara Bhat, J., Alva, V. D., 2011. A study of aluminium corrosion inhibition in acid medium by an antiemetic drug. *Trans. Indian Inst. Met.* 64, 377-384.
- [84]. Fouda, A. S., El-Ewady, G., Ali, A. H., 2017. Modazar as promising corrosion inhibitor of carbon steel in hydrochloric acid solution. *Green Chem. Lett. Rev.* 10(2), 88-100.
- [85]. Matter, H. A., Ayad, T. M., Dagane, F. A. J., 2023. The Corrosion Inhibition of The Carbon Steel structure and Oil Pipelines in 1M H₂SO₄ by Expired Chlorhexidine and Rhuvex, *Drugs. SJUOB* (1).
- [86]. Jebakumar Immanuel Edison, T., Sethuraman, M. G., 2013. Electrochemical investigation on adsorption of fluconazole at mild steel/HCl acid interface as corrosion inhibitor. *Int. Sch. Res. Notices*. <http://dx.doi.org/10.1155/2013/256086>
- [87]. Espinoza-Vázquez, A., Rodríguez-Gómez, F.J., Negrón-Silva, G.E., González-Olvera, R., Ángeles-Beltrán, D., Palomar-Pardavé, M., Miralrio, A. Castro, M., 2020. Fluconazole and fragments as corrosion inhibitors of API 5L X52 steel immersed in 1M HCl. *Corros. Sci.* 174, 108853.
- [88]. Obot, I. B. Obi-Egbedi, N. O., 2008. Fluconazole as an inhibitor for aluminium corrosion in 0.1 M HCl. *Colloids Surf.* 330(2-3), 207-212.
- [89]. Kashyap, S., Kumar, S., Ramasamy, K., Lim, S. M., Shah, S. A. A., Om, H., Narasimhan, B., 2018. Synthesis, biological evaluation and corrosion inhibition studies of transition metal complexes of Schiff base. *Chem. Cent. J.* 12, 1-10.
- [90]. Obot, I. B., Ebenso, E. E., Obi-Egbedi, N. O., Afolabi, A. S., Gasem, Z. M., 2012. Experimental and theoretical investigations of adsorption characteristics of itraconazole as green corrosion inhibitor at a mild steel/hydrochloric acid interface. *Interm. Chem. Res.* 38(8), 1761-1779.
- [91]. Ansari, K. R., Quraishi, M. A., Ebenso, E. E., 2013. Electrochemical and thermodynamic investigation of diclofenac sodium drug as a potential corrosion inhibitor for mild steel in hydrochloric acid. *Int. J. Electrochem. Sci.* 8(12), 12860-12873.
- [92]. Almashhadani, H. A., Alshujery, M. K., Khalil, M., Kadhem, M. M., Khadom, A. A., 2021. Corrosion inhibition behavior of expired diclofenac Sodium drug for Al 6061 alloy in aqueous media: Electrochemical, morphological, and theoretical investigations. *J. Mol. Liq.* 343, 117656.
- [93]. Al-abbassi, A., Matrood, E., Dnkm, A., Sallh, A., Izrik, R., Suliman, M. 2020. Study of the Behavior of Sodium Diclofenac Expired Drug as an Corrosion Inhibitor of Iron in Sulfuric Acid Using UV-VIS Spectrophotometry. *Int. J. Sci. Res. Sci. Eng. Technol.* 233-242.
- [94]. Singh, A., Ebenso, E. E., Quraishi, M. A., 2012. Theoretical and electrochemical studies of metformin as corrosion inhibitor for mild steel in hydrochloric acid solution. *Int. J. Electrochem. Sci.* 7(5), 4766-4779.
- [95]. Onyeachu, I. B., Abdel-Azeim, S., Chauhan, D. S., Quraishi, M. A., 2020. Electrochemical and computational insights on the application of expired Metformin drug as a novel inhibitor for the sweet corrosion of C1018 steel. *ACS omega*, 6(1), 65-76.
- [96]. Matter, H. A., Ayad, T. M., Dagane, F. A. J., 2023. The Corrosion Inhibition of The Carbon Steel structure and Oil Pipelines in 1M H₂SO₄ by Expired Chlorhexidine and Rhuvex, *Drugs. SJUOB* 36(1).
- [97]. Al-Mhyawi, S. R., 2022. Application of expired Tramadol medicinal drug for

- corrosion inhibition of steel in acidic environment: Analytical, kinetic, and thermodynamic studies. *Int. J. Corros. Scale Inhib.* 11(3), 1282-1302.
- [98]. Abdallah, M., Gad, E. A. M., Sobhi, M., Al-Fahemi, J. H., Alfakeer, M. M., 2019. Performance of tramadol drug as a safe inhibitor for aluminum corrosion in 1.0 M HCl solution and understanding mechanism of inhibition using DFT. *Egypt. J. Pet.* 28(2), 173-181.
- [99]. Hameed, R. A., 2011. Ranitidine drugs as non-toxic corrosion inhibitors for mild steel in hydrochloric acid medium. *Port Electrochem Acta.* 29(4), 273-285.
- [100]. Abdel Hameed, R. E. D. A., 2009. Expired Ranitidine Drugs as Corrosion Inhibitor for Corrosion of Aluminum in Hydrochloric Acid. *Al-Azhar Bull. Sci.* 20(1-A), 151-163.
- [101]. Abd El Aziz, S. F., El, A. A. M., 2018. Expired concor drug as potential nontoxic corrosion inhibitor for 304 stainless steel in hydrochloric acid solution. *Materials Protection.* 59(2), 226-236.
- [102]. Gupta, N. K., Gopal, C. S. A., Srivastava, V., Quraishi, M. A., 2017. Application of expired drugs in corrosion inhibition of mild steel. *Int. J. Pharm. Chem. Anal.* 4(1), 8-12.
- [103]. Yousif, Q. A., 2021. Environmentally sustainable disposal of expired albendazole medication and its role in the protection of mild steel. *IOP Conf. Ser.: Earth Environ. Sci.* 790, 012072.
- [104]. Beltran-Perez, C., Serrano, A. A., Solís-Rosas, G., Martínez-Jiménez, A., Orozco-Cruz, R., Espinoza-Vázquez, A., Miralrio, A., 2022. A general use QSAR-ARX model to predict the corrosion inhibition efficiency of drugs in terms of quantum mechanical descriptors and experimental comparison for lidocaine. *Int. J. Mol. Sci.* 23(9), 5086.
- [105]. Chaudhari, L. P., Patel, S. N., 2019. Corrosion inhibition study of expired acetazolamide on mild steel in dilute hydrochloric acid solution. *J. Bio Tribo-Corros.* 5, 1-13.
- [106]. Fouda, A. S., Etaiw, S. H., Wahba, A., 2015. Effect of acetazolamide drug as corrosion inhibitor for carbon steel in hydrochloric acid solution. *Nat Sci.* 13(9), 1-8.
- [107]. Feng, L., Zhang, S., Zhou, Y., Pan, R., Du, H., Liu, F., Yang, Y., 2023. Expired glucosamine drugs as green corrosion inhibitors for carbon steel in H₂SO₄ solution and synergistic effect of glucosamine molecules with iodide ions: combined experimental and theoretical investigations. *Crystals*, 13(2), 205.
- [108]. Karthik, G., Sundaravadivelu, M., 2016. Studies on the inhibition of mild steel corrosion in hydrochloric acid solution by atenolol drug. *Egypt. J. Pet.* 25(2), 183-191.
- [109]. Raghavendra, N., Hubikar, L. V., Ganiger, P. J., Bhinge, A. S., 2020. Prevention of aluminum corrosion in hydrochloric acid using expired oseltamivir drug as an inhibitor. *J. Fail. Anal. Prev.* 20(6), 1864-1874.
- [110]. Hameed, R.A., Aleid, G.M., Khaled, A., Mohammad, D., Aljuhani, E.H., Al-Mhyawi, S.R., Alshammary, F. Abdallah, M., 2022. Expired dulcolax drug as corrosion inhibitor for low carbon steel in acidic environment. *Int. J. Electrochem. Sci.* 17(6), 220655.
- [111]. Sharma, S., Ganjoo, R., Kr. Saha, S., Kang, N., Thakur, A., Assad, H., Sharma, V. Kumar, A., 2022. Experimental and theoretical analysis of baclofen as a potential corrosion inhibitor for mild steel surface in HCl medium. *J. Adhes. Sci. Technol.* 36(19), 2067-2092.
- [112]. Singh, P., Chauhan, D. S., Srivastava, K., Srivastava, V., Quraishi, M. A., 2017. Expired atorvastatin drug as corrosion inhibitor for mild steel in hydrochloric acid solution. *Int. J. Indus. Chem.* 8, 363-372.
- [113]. Aljourani, J., Raeissi, K., Golozar, M. A., 2009. Benzimidazole and its derivatives as corrosion inhibitors for mild steel in 1M HCl solution. *Corros. Sci.* 51(8), 1836-1843.
- [114]. Raghavendra, N., 2019. Expired amitriptyline drug as a new nontoxic inhibitor protecting mild steel corrosion in HCl solution. *Sci. Lett.* 7(1), 26-31.
- [115]. Wang, X., Guo, H., Cai, S., Xu, X., 2023. Expired antihypertensive drugs as eco-friendly and efficient corrosion inhibitors for carbon steel in CO₂-saturated oilfield water: Experimental and theoretical approaches. *J. Mol. Struct.* 1294, 136555.
- [116]. Fouda, A. A., Idress, A. A., Atia, M. F., Yassen, M. E., 2021. Candesartan: a pharmaceutical product as effective corrosion inhibitor for Aluminum in Acidic Solution. *Zašt. mater.* 62(3), 204-212.
- [117]. Kardas, G., Solmaz, R., 2006. Electrochemical investigation of barbiturates

- as green corrosion inhibitors for mild steel protection. *Corros. Rev.* 24(3-4), 151-172.
- [118]. Usman, B. J., Gasem, Z. M., Umoren, S. A., Solomon, M. M., 2019. Eco-friendly 2-Thiobarbituric acid as a corrosion inhibitor for API 5L X60 steel in simulated sweet oilfield environment: Electrochemical and surface analysis studies. *Scientific reports*, 9(1), 830.
- [119]. Feng, L., Zhang, S., Zhou, Y., Pan, R., Du, H., Liu, F., Yang, Y. 2023. Expired glucosamine drugs as green corrosion inhibitors for carbon steel in H₂SO₄ solution and synergistic effect of glucosamine molecules with iodide ions: combined experimental and theoretical investigations. *Crystals*. 13(2), 205.
- [120]. Fouda, A. S., Ibrahim, H., Rashwaan, S., El-Hossiany, A., Ahmed, R. M., 2018. Expired drug (pantoprazole sodium) as a corrosion inhibitor for high carbon steel in hydrochloric acid solution. *Int. J. Electrochem. Sci.* 13(7), 6327-6346.
- [121]. Al-Shafey, H. I., Hameed, R. A., Ali, F. A., Aboul-Magd, A. E. A. S., Salah, M., 2014. Effect of expired drugs as corrosion inhibitors for carbon steel in 1M HCL solution. *Int. J. Pharm. Sci. Rev. Res*, 27(1), 146-152.